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„	10	„	„	"reduction"	„ "action."

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21	20	from bottom	*for*	$H.OCH_2$	*read* $HOCH_2$.
21	20	„ „	„	H.OC	„ HOC.
24	10	„ „	„	Br_2CN	„ $NCBr_2$.
24	10	„ „	„	Br_2CNO	„ $(O_2N)CBr_3$.
24	16	„ „	„	$OCN\diagdown\atop\vert\quad\rangle$ $CN\diagup$	„ $OCN\diagdown\atop\vert\quad\rangle O.$ $CN\diagup$
25	8	from top,	„	(HO.N)C.H.NO)	*read* (HO.N)C.CH.NO.
26	bottom line.		The formula for fulminic acid should be		$\overset{N.C.OH}{\underset{C(N.OH).}{\diagdown\diagup}}$
26	5	from top,	*for*	HO.N : CH	*read* HO.N : C.OH.
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288	17	„ top,	„	"form of oxidation here assumed to take,"	*read* "assumption that such an oxidation takes."
408	3	„ „	„	"P. D. Hein,"	*read* "P. De Heen."

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